

Sirvan Sultan Uguz<sup>1,3</sup>, Baris Ozel<sup>1,2</sup>, Leonid Grunin<sup>3</sup> and Mecit Halil Oztop<sup>1</sup>

<sup>1</sup>Department of Food Engineering, Middle East Technical University, Ankara, Turkey

<sup>2</sup>Department of Food Engineering, Ahi Evran University, Kirsehir, Turkey

<sup>3</sup>Resonance Systems GmbH, Kirchheim unter Teck, Germany



## Abstract

Hard candies produced by sucrose and doctoring agents such as glucose syrup (GS) and high fructose corn syrup (FS) have been investigated in terms of their final composition, glass transition temperature ( $T_g$ ), total soluble solids (TSS) content and water activity ( $a_w$ ). Additionally, time domain (TD) 1H NMR parameters of longitudinal relaxation time ( $T_1$ ) and second moment ( $M_2$ ) measurements have been used to understand the glassy state and crystallization characteristics of different hard candy formulations. Prepared candies include sucrose as the main sugar component and different levels of doctoring agents have been mixed with sucrose to obtain products with different characteristics. Results showed that addition of any doctoring agent to sucrose formulations decreased the  $T_g$  of the system significantly ( $p \leq 0.05$ ). Furthermore, GS or FS addition also induced significant changes in TSS and  $a_w$  of the samples.  $T_1$  and  $M_2$  results were almost parallel to each other, both produced the highest values at the highest sucrose concentration ( $p \leq 0.05$ ). This study demonstrated that the glass transition and crystallization characteristics of hard candy formulations can be monitored and analyzed by TD NMR relaxometry, alternative to other frequently used conventional methods including differential scanning calorimetry (DSC) and X-ray diffraction.

## Materials & Methods



SpinTrack, Resonance Systems

NMR experiments were performed by using a instrument a 20.34 MHz 10 mm r.f. coil NMR system (Spin Track SB45, Kirchheim U.Teck, Germany)).

Saturation Recovery  $\rightarrow T_1$  (ms/s)

Magic Sandwich Echo  $\rightarrow M_2$

DSC  $\rightarrow T_g$



Water Activity



Differential Scanning Calorimetry



Total Soluble Solids (TSS)

## Sample Preparation



Mixing up the ingredients



Boiling up to 130°C



Pouring into moulds



Cooling to room temperature



Storing in sealed pans

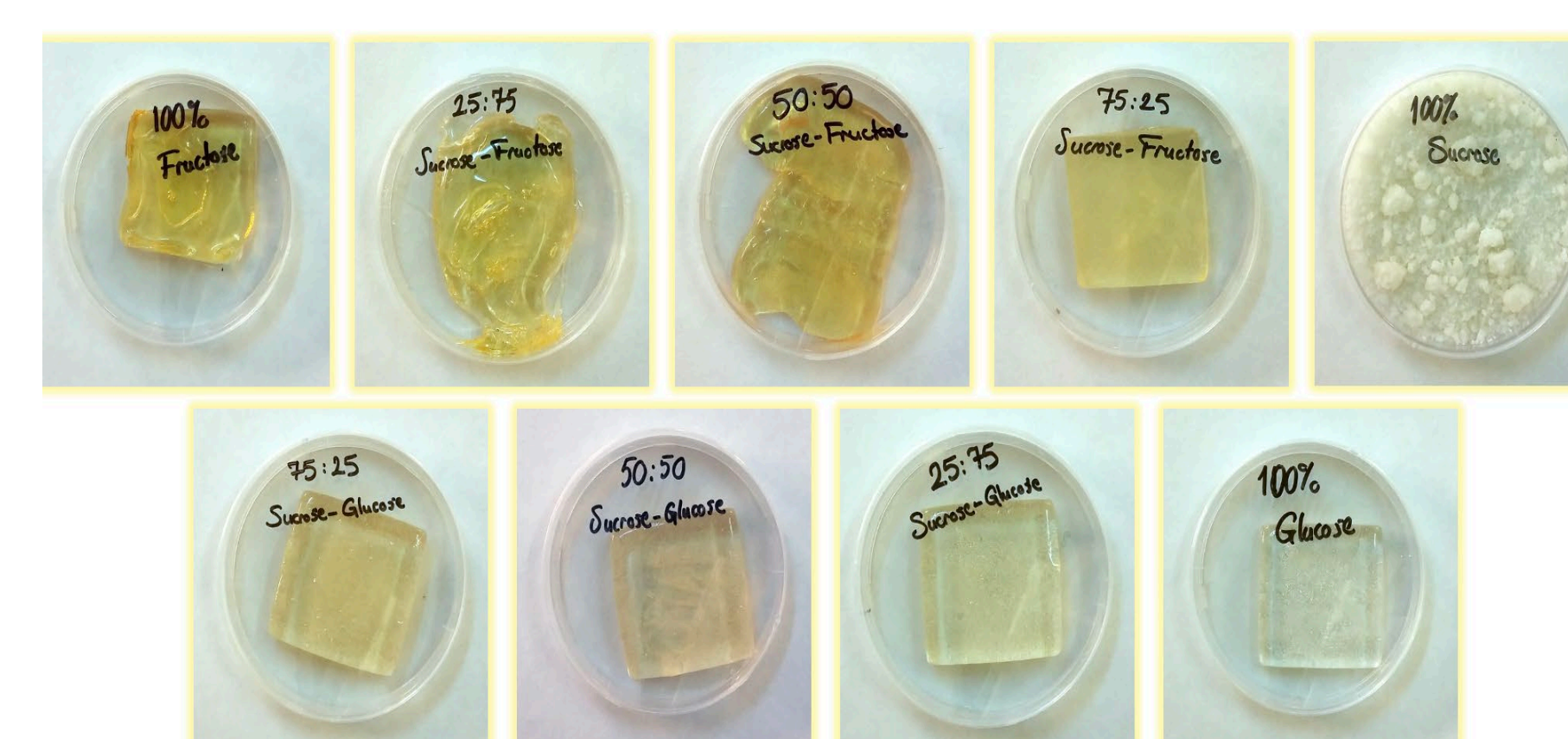


Figure 1. Hard Candies with different formulations

## Results

Table 1.  $T_g$ , TSS and  $a_w$  values of hard candies

Sample	$T_g$ (°C)	TSS (°Bx)	$a_w$
100:0_FS	2.13	92.04±0.12 <sup>d</sup>	0.41±0.01 <sup>c</sup>
75:25_FS	4.03	93.46±0.16 <sup>c</sup>	0.35±0.01 <sup>d</sup>
50:50_FS	15.21	92.59±0.16 <sup>d</sup>	0.41±0.02 <sup>c</sup>
25:75_FS	25.61	92.63±0.42 <sup>d</sup>	0.37±0.01 <sup>d</sup>
100_S	53.32	90.96±0.17 <sup>e</sup>	0.80±0.02 <sup>a</sup>
25:75_GS	34.30	94.26±0.12 <sup>b</sup>	0.42±0.01 <sup>c</sup>
50:50_GS	44.02	94.52±0.43 <sup>ab</sup>	0.48±0.01 <sup>b</sup>
75:25_GS	43.38	90.47±0.40 <sup>e</sup>	0.42±0.01 <sup>c</sup>
100:0_GS	45.23	95.07±0.17 <sup>a</sup>	0.43±0.01 <sup>c</sup>

- Highest  $a_w$  :100\_S. Addition of any doctoring agent decreased  $a_w$ . GS – sucrose formulations had higher  $a_w$  than the FS – sucrose. Addition of a doctoring agent increased the TSS. The increase in TSS was more evident with GS addition. GS – sucrose candies : higher TSS profile than FS – sucrose candies (Except for the 75:25\_GS formulation). 100\_S :the highest  $T_g$  as 53.32 °C . Addition of FS or GS produced lower  $T_g$  values. The  $T_g$  values of GS – sucrose candies were higher than that of FS. Although, GS addition to sucrose candies also decreased  $T_g$ , There is not a clear trend between GS concentration and  $T_g$  values. The  $T_g$  results showed that the doctoring agents (FS and GS) clearly decreased sucrose crystallization taking place in the rubbery state during the cooling period.

Table 2.  $T_1$  and  $M_2$  values of hard candies

Sample	$T_1$ (ms)	$M_2$
100:0_FS	101.75±6.43 <sup>f</sup>	6.85±0.08 <sup>e</sup>
75:25_FS	138.95±0.64 <sup>c</sup>	8.71±0.06 <sup>c</sup>
50:50_FS	111.67±0.32 <sup>e</sup>	8.07±0.04 <sup>d</sup>
25:75_FS	121.80±1.99 <sup>d</sup>	8.91±0.13 <sup>c</sup>
100_S	1627.00±22.61 <sup>a</sup>	10.96±0.19 <sup>a</sup>
25:75_GS	115.50±0.53 <sup>e</sup>	8.76±0.12 <sup>c</sup>
50:50_GS	149.20±0.14 <sup>b</sup>	9.38±0.09 <sup>b</sup>
75:25_GS	68.70±1.21 <sup>g</sup>	5.26±0.17 <sup>f</sup>
100:0_GS	123.05±0.78 <sup>d</sup>	9.01±0.07 <sup>c</sup>

- $T_1$  has monoexponential relaxation behavior except for 100\_S. The relaxation peaks of 100\_S candies exerted a very long  $T_1$  (1627 ms) and a much shorter (~78 ms). Generally, longer  $T_1$  values are observed with compact and highly ordered crystalline state arrangements at high solid concentrations.100\_S samples having the highest  $T_g$  also attained the longest  $T_1$  which was compatible with the claim that related the long  $T_1$  with the higher order of the crystalline state. Addition of doctoring agents extremely lowered the  $T_1$  values in consistence with their sucrose crystallization retarding effect.
- $M_2$  can be used to understand crystalline state molecular dynamics. A higher  $M_2$  value is associated with a lower proton mobility in a solid state. 100\_S candies had the highest  $M_2$  (10.96) means highest crystallinity level. Similar to  $T_1$  results, incorporation of doctoring agents also reduced the  $M_2$ , thus the degree of crystallinity of the sucrose hard candies

## References

1. Grunin, L.; Oztop, M.H.; Guner, S.; Baltaci, S.F. Exploring the crystallinity of different powder sugars through solid echo and magic sandwich echo sequences. Magn. Reson. Chem. 2019, 57, 607–615. doi:10.1002/mrc.4866.
2. Hartel, R.W.; Ergun, R.; Vogel, S. Phase/State Transitions of Confectionery Sweeteners: Thermodynamic and Kinetic Aspects. Compr. Rev. Food Sci. Food Saf. 2011, 10, 17–32.



Funded by the European Union

This study has received funding from the European Union's Horizon 2020 Research and Innovation Programme under grant agreement 101008228. Results reflects only the author's view, and that the Agency is not responsible for any use that may be made of the information it contains.